Hot Science Cool Talks

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#31

What's in the Water? The History and Future of Barton Springs

Dr. Barbara Mahler September 10, 2004

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What's in the Water? The history and future of Barton Springs

Dr. Barbara Mahler







Karst and Karst Aquifers

- Karst is a distinctive type of landscape that is formed from the dissolution of soluble rocks, including limestone and dolomite
- Karst is characterized by sinkholes and caves and underground drainage systems
 - Karst aquifers, that is, aquifers in karst regions, are capable of supplying large quantities of water
 - Karst landscapes are often very scenic areas

Karst aquifers are different from other aquifers (such as sandstone) in that:

- Water doesn't flow similarly in all parts of the aquifer (Inhomogeneous)
- Water flows preferentially in one direction rather than another (Anisotropic)



Sandstone aquifer

Water velocities are a few inches per day

Water velocities can be miles per day

Karst aquifers are very productive but very vulnerable to contamination

Flow through solution "pipes"



Thin soils

Urbanization: How has Austin grown since 1839



We live here because of karst ...





East





Barton Creek (downstream)

Old Mill Spring

Eliza Spring

Barton Creek pool bypass

Main Barton Spring

Upper Barton Spring









What can we measure in the water?

- "Physical" parameters
- Major ions (calcium, sodium, etc.)
- Nutrients
- Contaminants in water
 - Pesticides, VOCs, pharmaceuticals, waste-water indicators
- Contaminants on sediment
 - Metals, PAHs, organochlorine compounds



Physical parameters can be measured continuously

- Discharge (flow rate)
- Temperature
- Conductivity
- Turbidity
- pH
- Dissolved oxygen





Baseflow: physical parameters change slowly as the aquifer drains

Barton Springs, July 2004



... but when it rains, things happen!



- Velocity/Springflow changes reflect aquifer and local conditions
- Conductivity reflects rainfall infiltrating
- Turbidity corresponds to turbid creek recharge

Springflow and surface-water flow are closely connected



Major Ions: the natural chemical signature of the Water

Cations

Ca, Mg, Sr, Na

Anions

HCO₃, SO₄, CI

Help us differentiate between different sources of water feeding the springs

Sodium, baseline



Sodium, during pool drawdown





Nutrients: The Right Balance Needed

- Nitrogen, phosphorous necessary for plant growth
- Additional sources: fertilizer, wastewater, animal wastes
- Too much causes eutrophication

NO₂ + NO₃ August and September 2003



- "New" (post-DDT) pesticides are very soluble
- Detected in 99% of surface water and 49% of ground water sampled by the USGS
- Very low detection limit
- Toxic by design



Dissolved Contaminants: Pesticides



After rain, pesticide concentrations increase

Occurrence of Atrazine in response to May 1, 2000 rain even



Dissolved contaminants: Volatile organic compounds (VOCs)



Gasoline compounds - BTEX - MTBE (oxygenate) Cleaning solvents (degreasers) - Industrial processes - Dry cleaners - Floor cleaners

Dissolved contaminants: "Emerging contaminants"

Pharmaceuticals

- Excreted into waste water
- Antibiotics, hormones, analgesics

 Other waste-water indicators
 Flame retardants, surfactants, fragrances

Planned for future sampling









What's in the sediment?



Toxic chemicals taint Barton waters

TOXIC WATERS

Decades-old fuel waste cited as possible source

City closes Barton pool

POOL, OTHER CITY CREEKS MAY POSE HEALTH RISK

Suspended-sediment sampling

- Metals

 Zinc, lead, mercury

 Organochlorine compounds

 PCBs, DDT, chlordane

 PAHs
 Compare between urban
 - creeks and spring sediment



Comparing metals in springs and creeks



- Some metals (As, Ni, Cr, Cd) have higher concentrations in the springs
- High concentrations in creeks are associated with the natural geochemistry of aquifer sediment

Comparing metals in springs and creeks



- Two metals (Pb, Zn) have higher concentrations in the creeks
- High concentrations
 of Pb and Zn in creeks
 are associated with
 urbanization



DDE and PAH in spring and creek sediments

- Either not detected or very low in springs
- PAHs are higher than sediment quality guidelines in urban stream suspended sediments

Polycyclic Aromatic Hydrocarbons (PAH) in the urban environment

- Formed from 2 to 6 fused benzene rings
- Largest group of suspected carcinogens
- Formed by combustion of organic matter



So where are all those PAHs coming from?

A. B. A. MALO

What's in them? Coal-tar emulsion vs. asphalt-emulsion

Coal tar - final product of destructive distillation of coal to produce coke and (or) gas

Coal tar contains 50% or more PAHs by weight

Coal tar is a known carcinogen & classified as a hazardous waste

Coal-tar based sealants are typically 20-35% coal tar

Coal-tar based sealants are thought to be more widely used in the east Asphalt - derived from refining of crude petroleum

Asphalt contains PAHs at concentrations orders of magnitude less than coal tar

Asphalt-based sealants are typically 28-45% asphalt resin

Asphalt-based sealants are thought to be more widely used in the west

Sealant Product Analysis

(Data from City of Austin)



- 13 parkings lots were washed off
- Runoff filtered in the laboratory
 - Stainless-steel filter holder
 - 0.45 μ m Teflon filter
- Particles massaged from filter and analyzed







Particles washed off sealed parking lots are highly contaminated with PAH









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Dr. Barbara Mahler is a Research Hydrologist with the Water Resources Division of the U.S. Geological Survey. Dr. Mahler's research focus is on karst hydrogeology and sediment-associated contaminant transport. Over the past five years, she has co-authored 11 articles in peer-reviewed publications, co-authored 11 technical reports, and presented research results at numerous national and international meetings. Dr. Mahler received her Ph.D. from the University of Texas at Austin; she was a NSF-NATO International Postdoctoral Fellow, spending one year in Montpellier, France, investigating bacterial transport in karst.