Hot Science Cool Talks

UT Environmental Science Institute

5

True Gems: Origins and Identification

Dr. Mark Helper May 5, 2000

Produced by and for *Hot Science - Cool Talks* by the Environmental Science Institute. We request that the use of these materials include an acknowledgement of the presenter and *Hot Science - Cool Talks* by the Environmental Science Institute at UT Austin. We hope you find these materials educational and enjoyable.

Common questions:

- How are gems different from other minerals?
- How are gems identified?
- Where do they come from and why are they rare?
- How do man-made gems compare to naturals?



What is a gem?

- A natural material that is prized for:
 - Beauty
 - Durability
 - Rarity



Ametrine Quartz, Bolivia

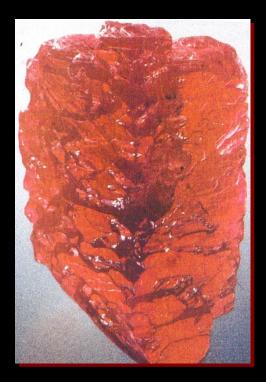
Gems are Beautiful

- Outstanding visual qualities:
- Color
- Brilliance
- Unusual optical phenomena
 - Stars, cat's eyes
 - Play-of-color

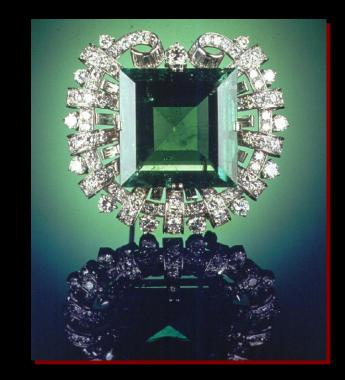


Rose Quartz, Brazil

Beauty: Color







Hixon Ruby

Logan Sapphire

Hooker Emerald

Beauty: Brilliance

- Clarity
- Scintillation (sparkle)
- "Fire" prism effect



Synthetic Rutile

Beauty: Unusual Phenomena

 Reflection of light from something within a gem or diffraction of light by the gem.





Opal

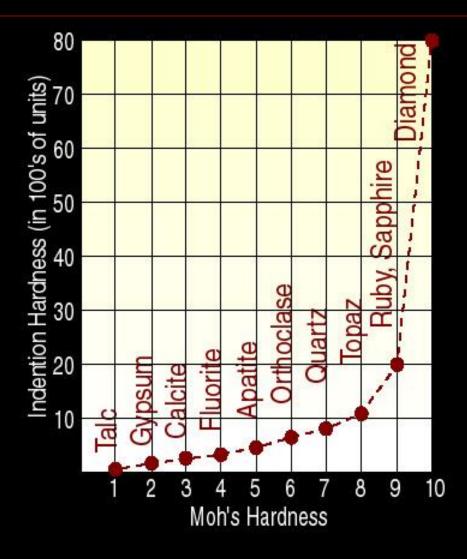
Cat's Eye Gemstones

Gems are Durable

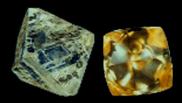
- Not easily scratched Hard
- Not easily broken Tough



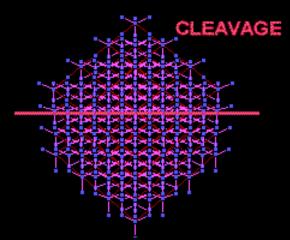
Cleaving a diamond



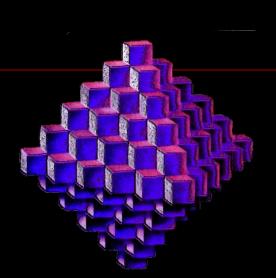
Durability: Cleavage in diamond



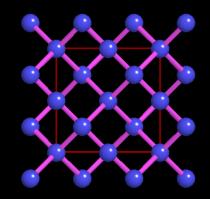
Diamond crystals



Planes of fewest bonds are cleavage



Stacked cubes form an octahedron



Cube of carbon atoms in diamond

Gems are Rare

Minerals with properties suitable for gems are rare
Gem quality specimens are even rarer









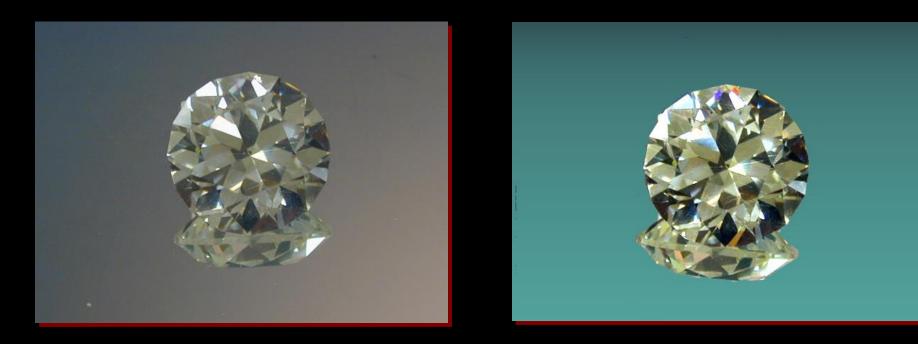
Emerald



How are gems identified?

- By measuring or observing properties:
 - Hardness
 - Specific Gravity
 - Interactions with light
- Microscopic observation of crystal growth features
- Special techniques

Gem Identification: Diamond, C.Z. & Specific Gravity



10 mm, 4.1 carats

10 mm, 8.0 carats

Diamond

Cubic Zirconia

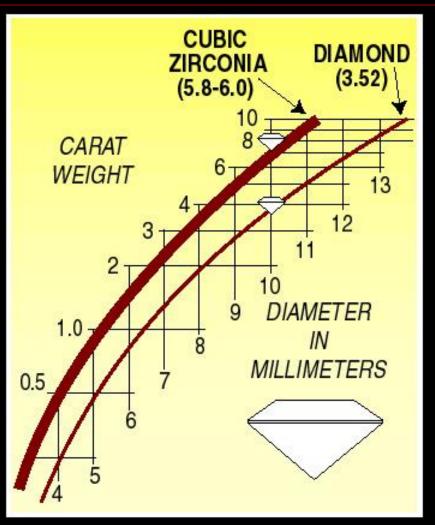
Gem Identification: Specific gravity curves



10 mm, 8.0 cts



10 mm, 4.1 cts

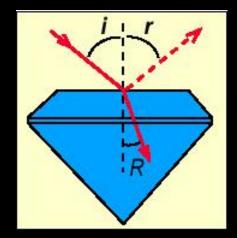


Gem Identification: Interactions with light

- Refraction bending
- Dispersion separation into colors
- Polarization filtering
- Absorption produces color

Interactions with light: **Refraction**

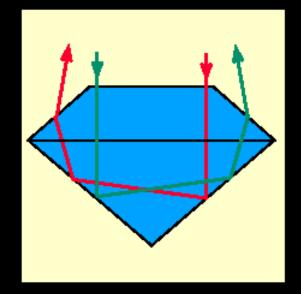
- The speed of light in gems is less than it is in air
- Decrease in speed results in bending of light rays = refraction
- Measure the amount of bending of light to identify gems

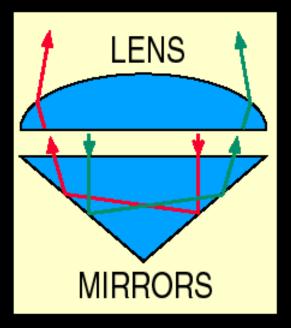


Interactions with light: How slow is light in gems?

		Refractive Index
	<u>V (km / sec)</u>	V air / V material
Diamond	124,083	2.42
Ruby	169,429	1.77
Emerald	189,803	1.58
Glass	197,349	1.52
Water	225,442	1.33
Air	299,890	1.00

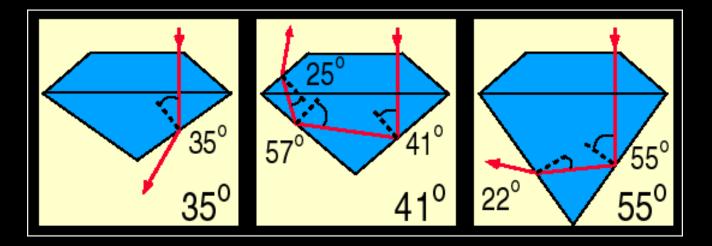
Interactions with light: Refraction and brilliance





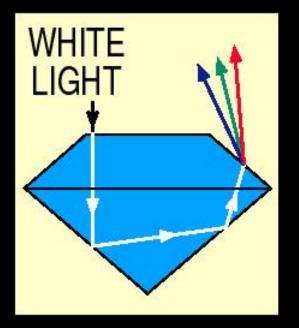
Interactions with light: Critical angle and brilliance

• For quartz, with a critical angle of 40° :



Interactions with light: **Dispersion**

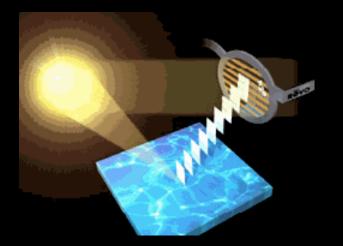
 Colors of the spectrum are separated by different amounts of refraction.

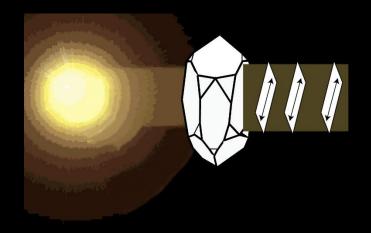




Interactions with light: Polarization

- Light passing through some minerals emerges as polarized light
- Test whether gem polarizes light





Light polarized by reflection

Light polarized by a mineral

Interactions with light: Absorption/Color

Not a distinctive property for most gem minerals



Ruby

Orange Sapphire

Sapphire

 $CORUNDUM - Al_2O_3$

Gem Identification: Under the Microscope

Natural



100X

Emerald

Opal

Ruby

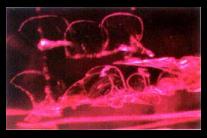
Man-made



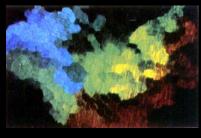
40X



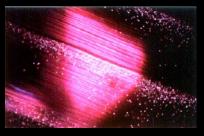
20X



45X



40X



60X

Why are gems rare?

- Contain rare chemical elements
- Get to the surface by unique sequences of geologic events
- Some require unusual conditions for formation

Gem Rarity: Rare chemical elements

- Emerald Beryllium & Chromium
- Ruby Chromium
- Sapphire Titanium
- Tourmaline Boron
- Topaz Fluorine



Star Ruby

Emerald

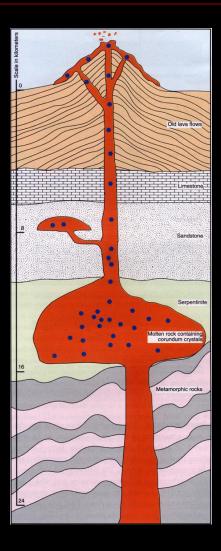




Gem Rarity: Transport to the surface

- Many gems form at great depths; need geologic process(es) to get them to the surface
 - Volcanic eruptions
 - Sapphire
 - Diamond
 - Mountain building processes
 - Ruby
 - Emerald
 - Many others

Transport to the surface: Volcanic eruptions

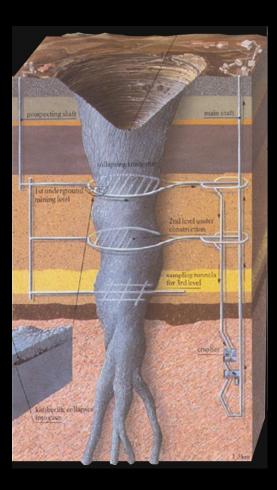




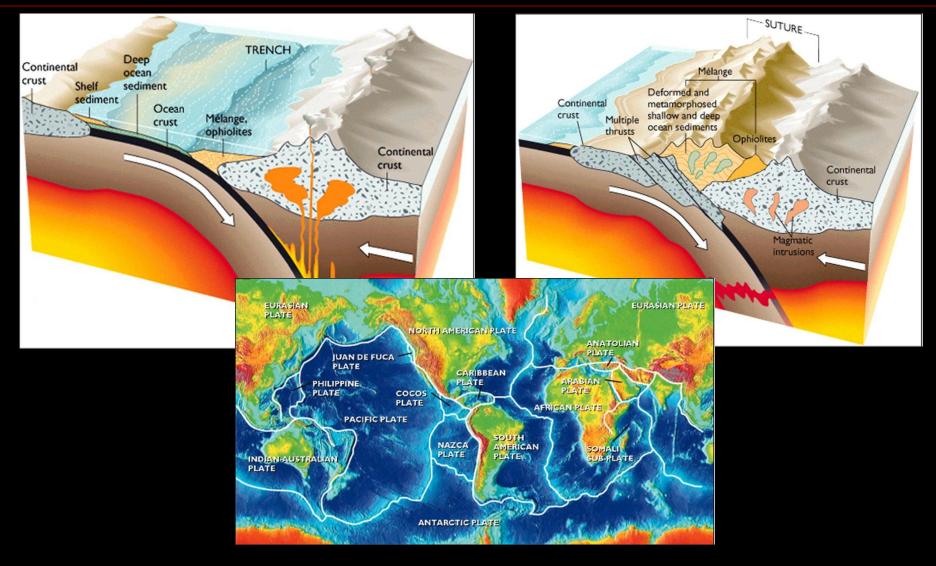
Montana Sapphire



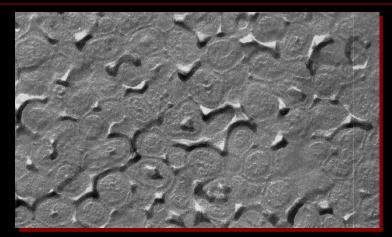
Diamond in kimberlite



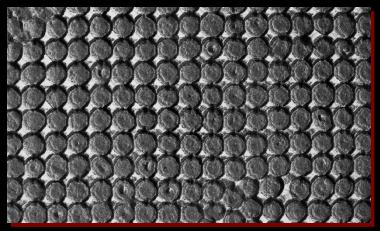
Transport to the surface: Mountain building



Gem Rarity: Unique processes of formation



Common Opal at 50,000 X



Precious Opal at 50,000 X



Where do gems come from?

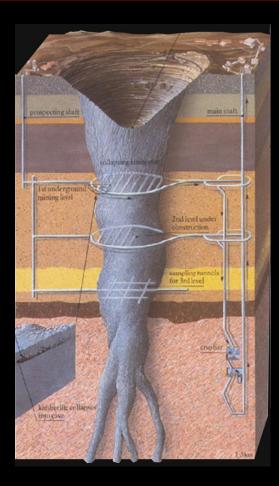


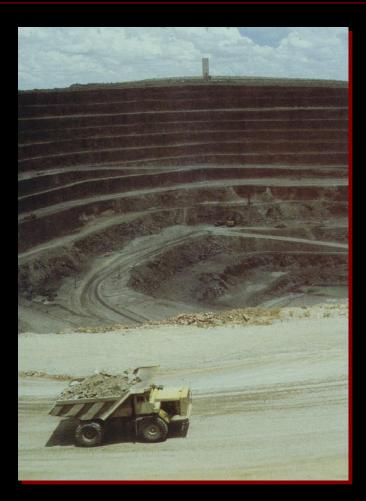
Lode Deposits: Muzo Emerald Mine, Colombia





Lode Deposits: Diamond Mining

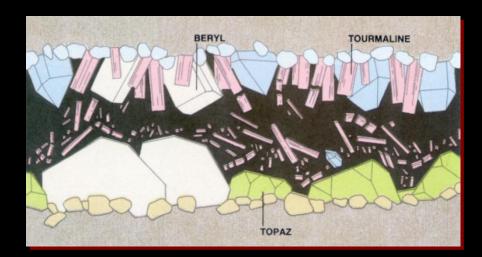




Lode Deposits: Pegmatites

- Unusual, small bodies of igneous rocks containing large crystals
- Sometimes rich in rare elements and minerals





Lode Deposits: Pegmatites

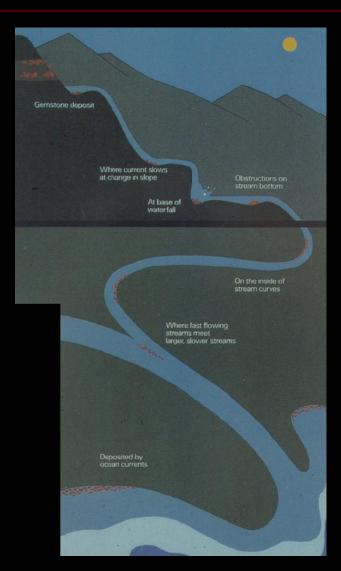


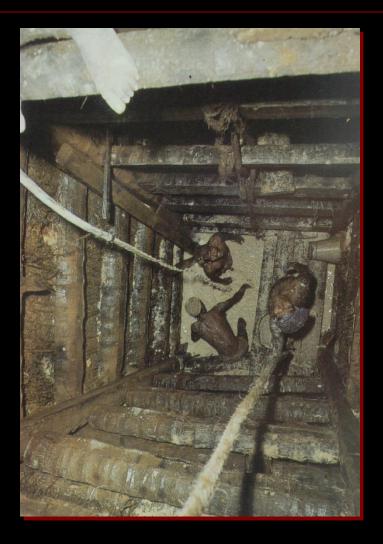






Gem Gravel Mining, Sri Lanka





Placer Deposits: Gem Gravel Mining, Sri Lanka

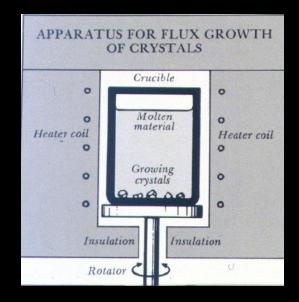




Man-Made Minerals as Gems

- Technological marvels, beautiful, but lack rarity
- Lab processes do not imitate nature





Ramaura Ruby

Summary

 How are gems different from other minerals?



Beautiful, Rare, Durable

- How are gems identified?
 - Physical and Optical Properties
 - Microscopic growth features

Summary

- Why are they rare and where do they come from?
 - Rare chemistry



- Unique processes of formation and/or transport to the surface
- Lode and placer deposits
- How do man-made gems compare to naturals?
 - Some are in every way identical, but lack rarity

Dr. Mark Helper

Mark Helper has degrees in Geology from the University of Illinois (BS) and University of Texas at Austin (Ph.D.). As a senior lecturer, he currently teaches classes in Field Geology and Gems and Gem Minerals, serves as director of the Department's summer field geology program, and is curator of the E. M. Barron and G. & M. Vargas gem and mineral collections. His research interests in mountain building processes have led to field studies of portions of the Klamath Mountains in northern California, the Llano uplift of central Texas, the Picuris Mountains of northern New Mexico, and the Shackleton Range and Heimefrontfiella of Antarctica. He has received the Knebel Award for excellence in undergraduate teaching (1995), the Antarctic Service Medal of the U.S. Antarctic Program (1995), the American Federation of Mineralogical Societies Honorary Scholarship Award (1996), and the Miningco.com Award for best World Wide Web mineralogy site (1998).

